

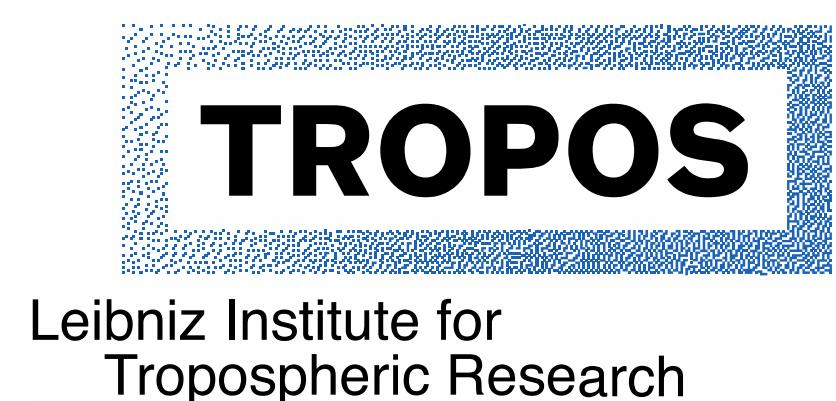
Results from the First CIMS Intercomparison Workshop at TROPOS in ACTRIS CiGas

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Introduction

- Secondary organic aerosols (SOA) have strong effects on Earth's radiation budget, local air quality and human health
- Recent progress in understanding SOA formation mechanisms through condensable vapors
- New instrumentation delivers detailed molecular information of oxidized organic compounds
- Specifically CIMS online instruments are widely used
- Comparability is largely unknown
- ACTRIS research infrastructure aims at better comparability
- ACTRIS CiGas focuses on condensable vapors

Experiments at the Atmospheric Chemistry Department Chamber ACD-C

- FEP twin-chamber setup
- V = 19 m³
- S/V ratio ~2.1
- Actively stirred T-controlled enclosure
- Continuous mode operation
- Set of UVA lamps
- Special glass inlet tube designed to ensure sampling of same composition and sampling length

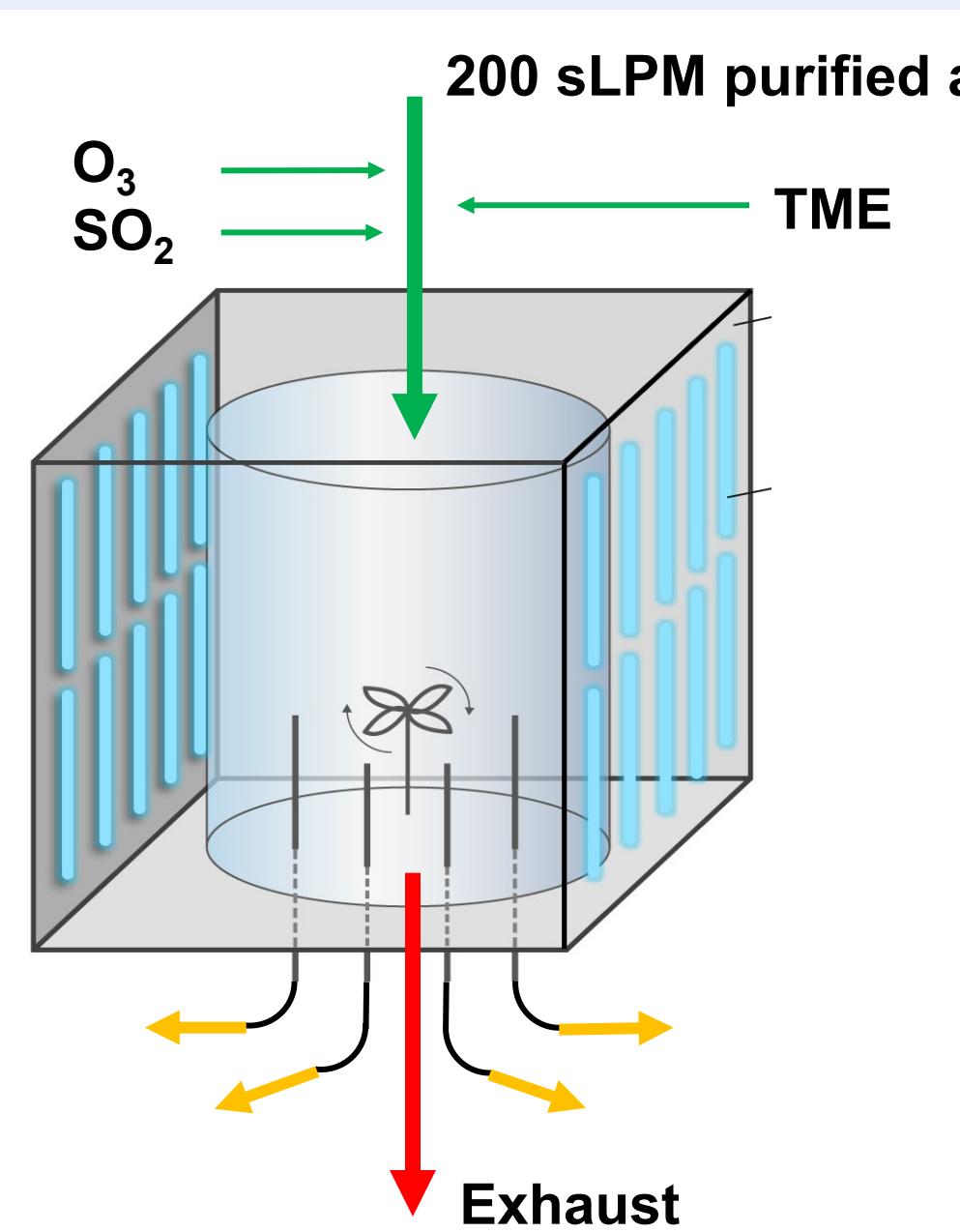


Fig. 1: Schematic experimental setup for *in-situ* sulfuric acid formation.

- Calibration experiments with 1,2-ISOPPOOH and 4-nitrophenol
- α -pinene oxidation experiments
 - OH oxidation
 - O₃ oxidation
 - O₃ oxidation + NO
- in-situ* sulfuric acid acid (SA) formation
 - 20 ppbv SO₂ + 5 ppbv O₃
 - Stepwise increase of TME
- Increase of relative humidity (RH) after each exp.
- Off-chamber activities
 - MS tuning at steady state
 - Calibration with sulfuric acid
 - Data analysis intercomparison

Instruments

10 instruments (5 connected to each chamber)

- Mediterranean Center for environmental Science - CEAM
- Helsinki University ACTRIS - UHEL
- TROPOS – TROPOS_ACDC
- Cyprus Institute - CYI
- Forschungszentrum Jülich – SAPHIR
- Goethe University Frankfurt - UFRA
- Goethe University Frankfurt - SCORPION
- Tofwerk - TOFWERK
- Tampere University - TAU
- TROPOS – TROPOS_MEL
- Different ToFMS units (**cToF**; **hToF**; **LToF**)
- Different primary ion sources (**corona discharge**, **X-ray**; **Am²⁴¹**; **Po²¹⁰**)
- Different inlet tube diameters and flow rates (1", 3/4", 1/2"; 2-23 sLPM)
- Different inlet designs (**Eisele**, **MION**, **AIM**, **SCORPION**, **FIGAERO**)

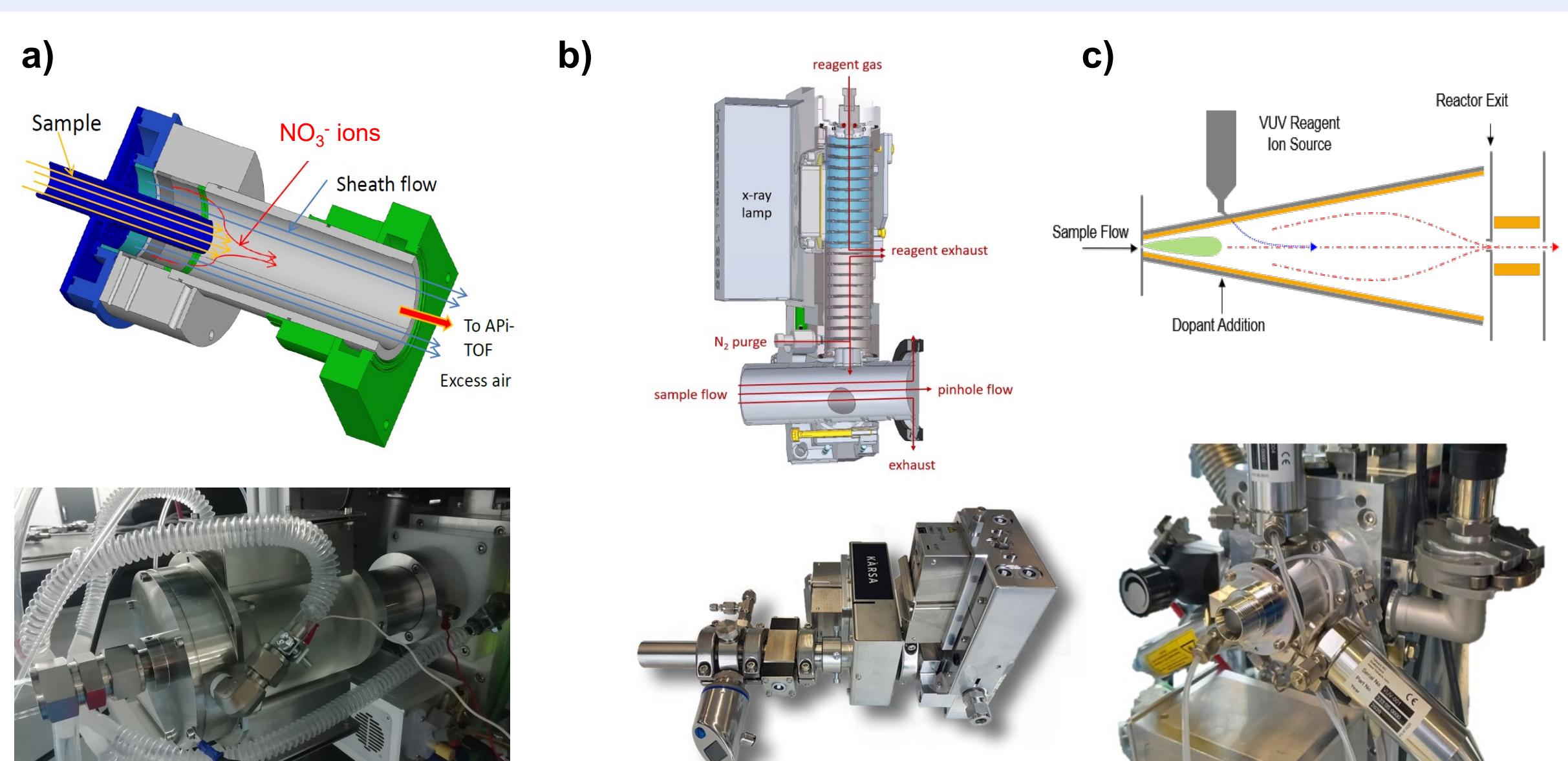


Fig. 2: Selected inlet designs of compared NO₃-CIMS instruments: a) Eisele-type inlets^[1-2] b) Multi-scheme chemical ionization inlet (MION)^[3] c) AIM reactor^[4].

Large influence of observed signals caused by variation of:

- Primary ion concentration
- Reaction chamber pressure
- Inlet line loss

Key results

- SA concentration** for steady states calculated from off-chamber **calibration**
- Mostly good qualitative agreement
- Differences in background level, absolute values
- Still **high variation** of SA concentrations of sulfuric acid, **despite thorough correction**
- Further discussion points:
 - Inlet tube effect (bends)
 - calibration setup differences
 - Background handling
- HOMs show larger variation in **SVOCs** and **ULVOCs** concentrations
- Comparatively good agreement in **LVOCs** and **ELVOCs**

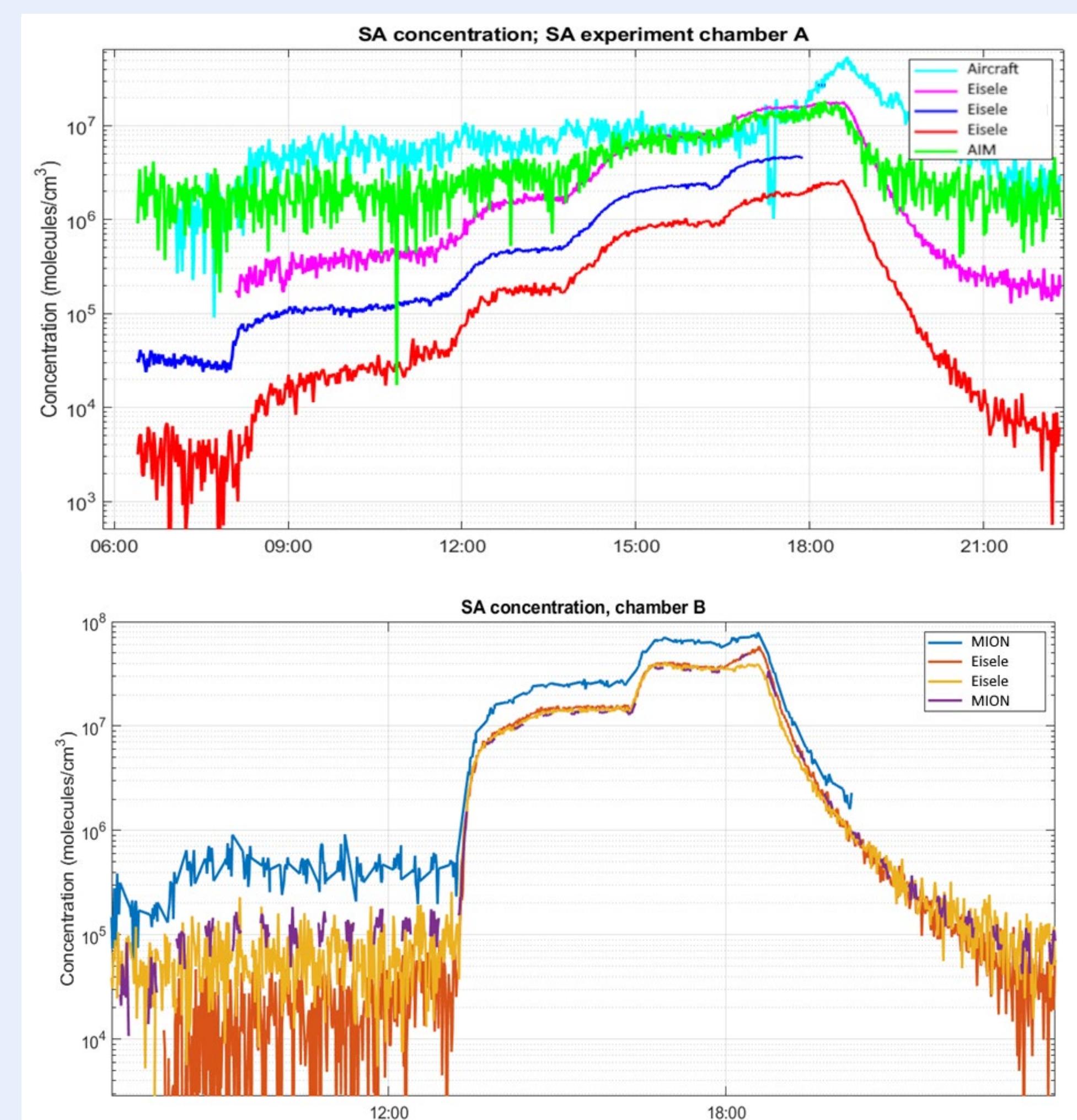


Fig. 3: Time series of inlet line loss corrected sulfuric acid concentrations for instruments in both chambers (chamber A upper plot; chamber B lower plot).

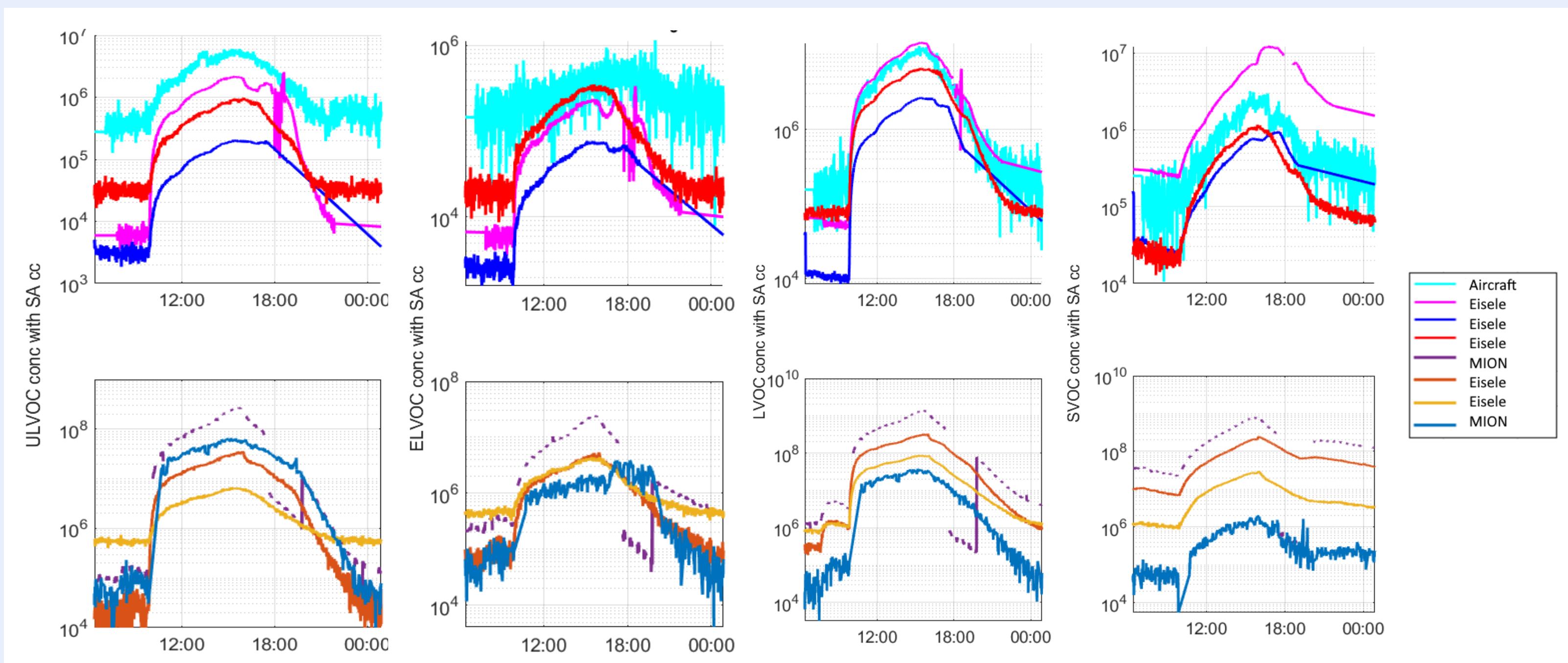


Fig. 4: Time series of inlet line loss corrected concentrations of HOMs classified by volatility^[5] for instruments in both chambers (chamber A upper plots; chamber B lower plots). Sulfuric acid concentrations were used to correct all CIMS data accordingly.

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