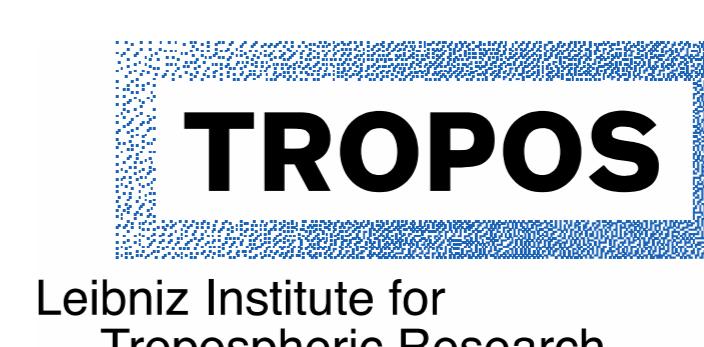


# Quantification of imidazoles in ambient aerosol particles from different environments in Europe and China

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## Motivation

Imidazoles are potential:

- Brown carbon constituents: They were found in numerous laboratory studies, as a secondary product of the reaction of glyoxal or methylglyoxal with nitrogen-containing compounds like ammonia, amines and amino acids (De Haan et al., 2009; Kampf et al., 2012; Powelson et al., 2014)
- Photosensitizers: They can convert volatile organic compounds to less volatile compounds leading to particle growth, a modified particle composition and thus modified physical properties as well as gas-phase HO<sub>2</sub> production (Aregahegn et al., 2013; González Palacios et al., 2016)
- Hazardous components of aerosol particles: Imidazoles, such as 4-methylimidazole, are known for their carcinogenic potential (National Toxicology, 2007)

However only few studies yielded qualitative evidence of imidazoles in ambient aerosol particles. Therefore, the aim was to detect and for the first time quantify imidazoles in ambient aerosol particles.

## Qualitative identification of imidazoles

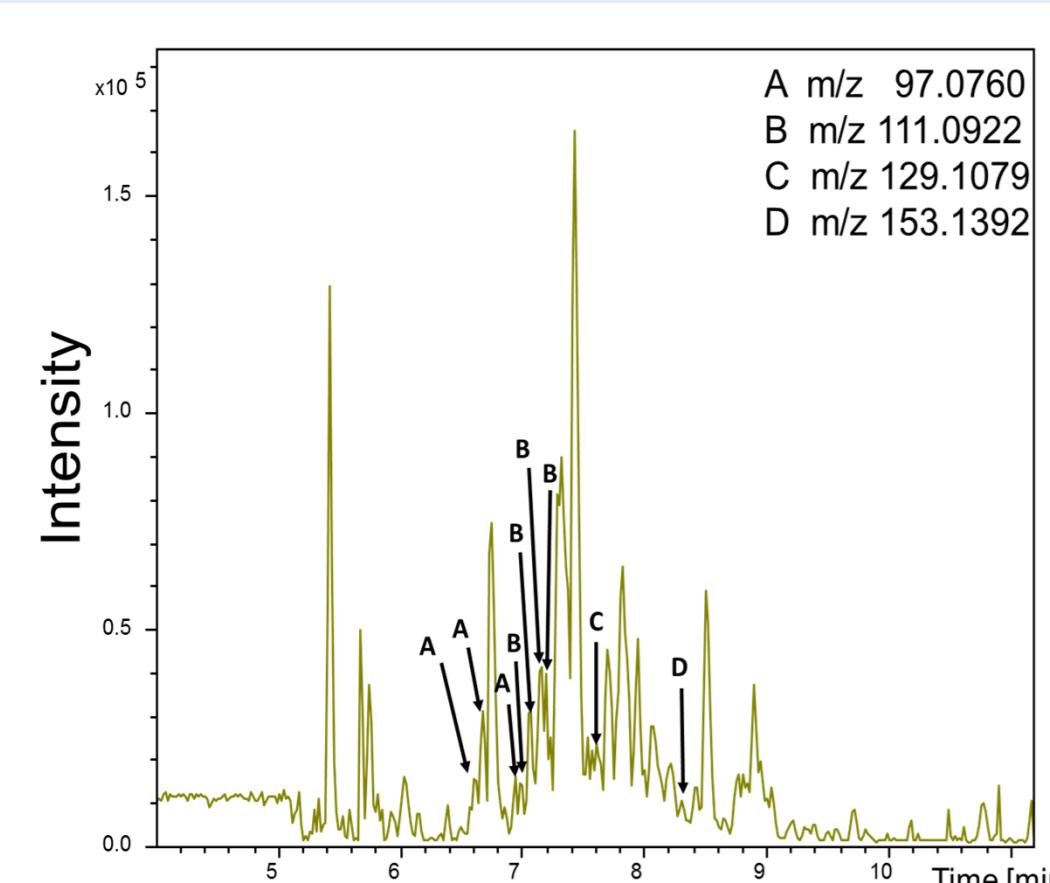


Figure 1. Base peak electropherogram obtained from the Melpitz sample of 16 July 2013. Base peaks referring to possible imidazole compounds are indicated with capital letters.

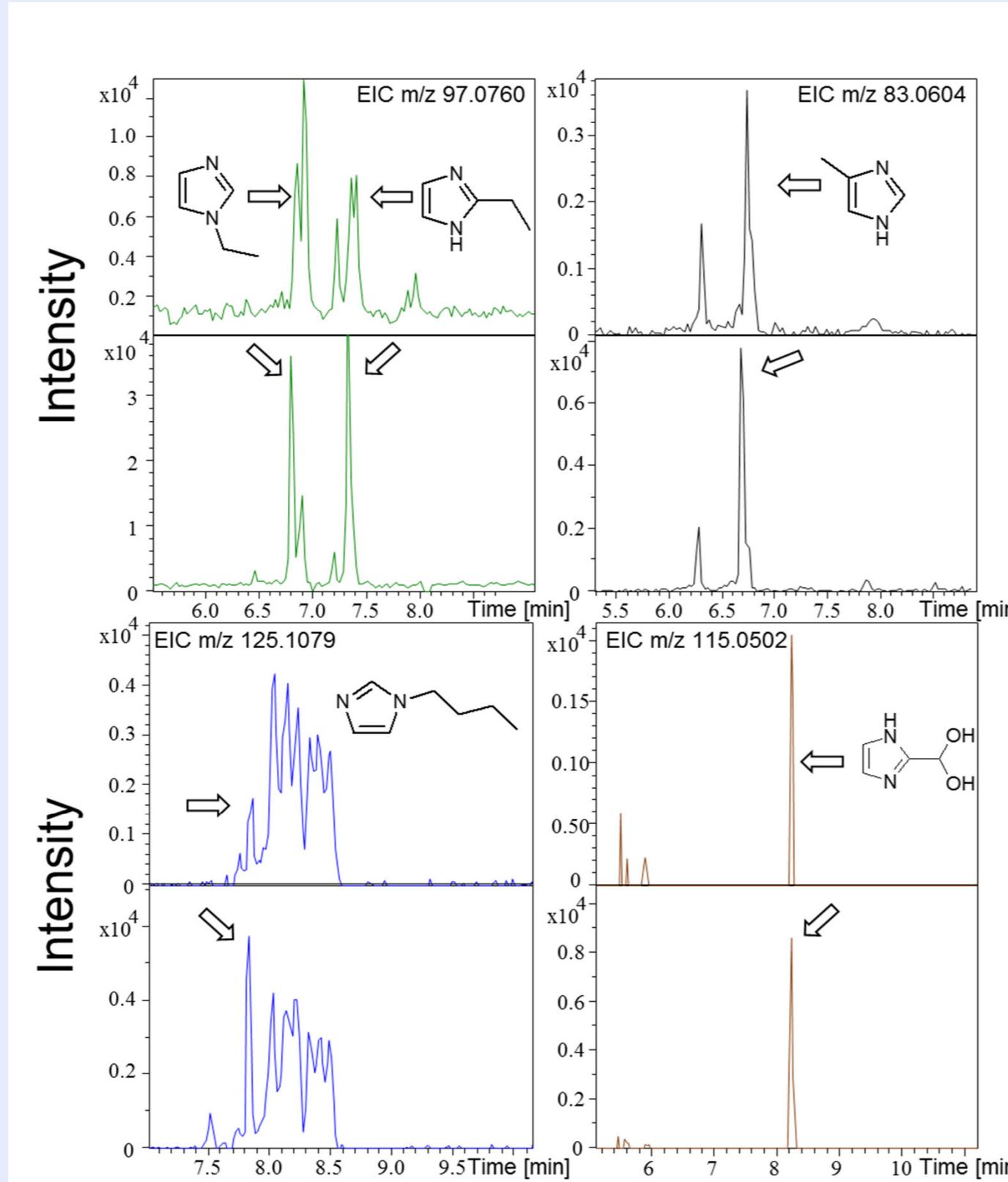


Figure 2. Selected extracted ion electropherograms of identified imidazoles after sample preparation of an ambient filter sample. Top: pure extract. Bottom: extract spiked with 5  $\mu$ M standard solution. Arrows mark the peak of the identified compound.

Identification of five imidazoles:

- 4-Methylimidazole (4MI)
- 1-Ethylimidazole (1EIM)
- 2-Ethylimidazole (2EIM)
- 1-Butylimidazole (BIM)
- Imidazole-2-carboxaldehyde (IC) in its hydrated form (HIC)

Evidences for the presence of a homologous series of alkyl-substituted imidazoles

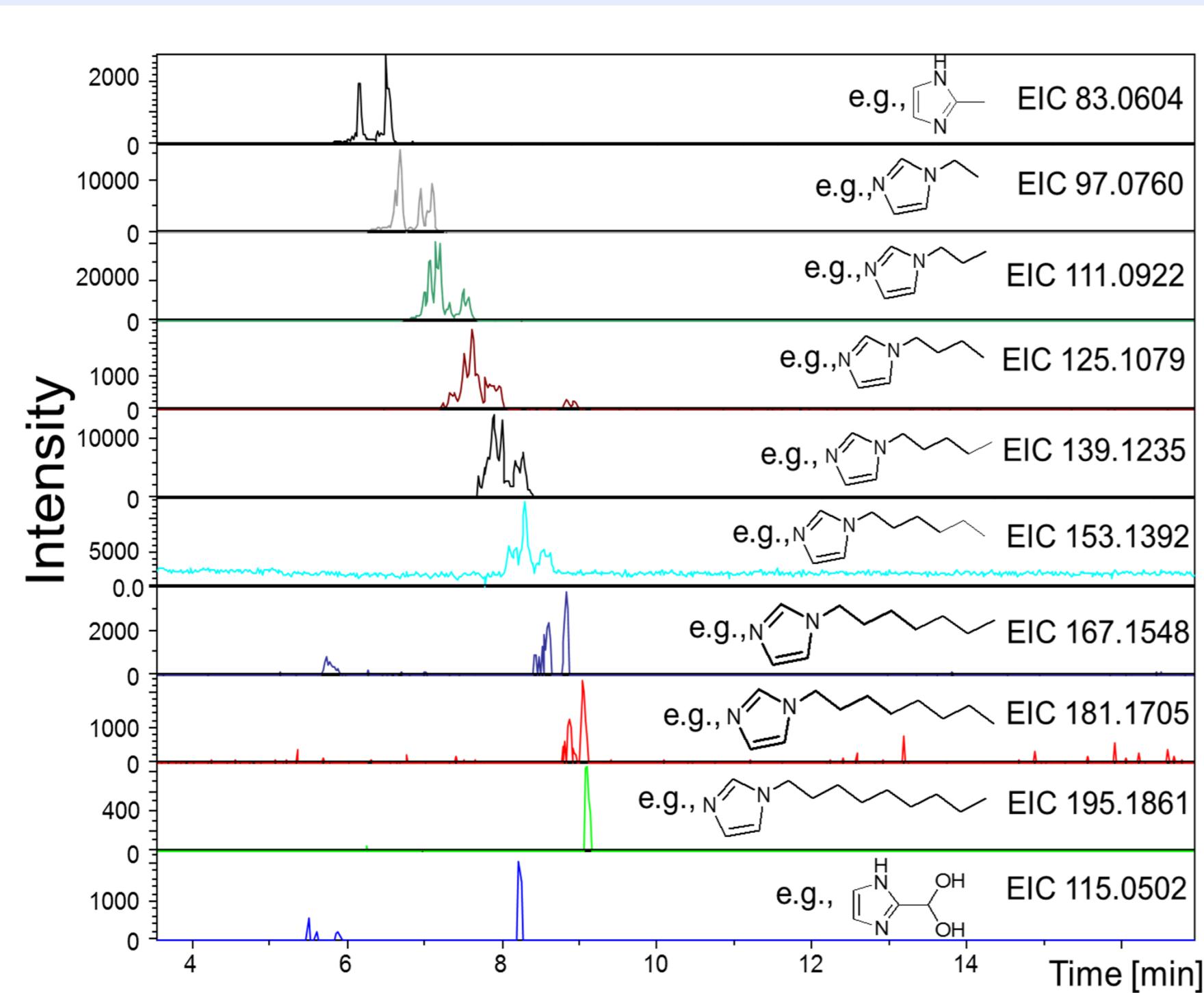


Figure 3. Extracted ion electropherograms of m/z values found in the literature and present study that are tentatively assigned to imidazoles, based on their exact mass. For each electropherogram an example for a possible structure is given.

## Conclusions

- First quantification of imidazoles in ambient aerosol particles
  - Signals corresponding to a homologous series of alkylated imidazoles were found
- This work:
- Corroborates laboratory studies by showing that imidazoles are present in ambient aerosol samples in measurable amounts.
  - Further motivates to explore the potential photosensitizing properties of small alkyl-substituted imidazoles.

## References

- De Haan, D. O.; Tolbert, M. A.; Jimenez, J. L. Atmospheric condensed-phase reactions of glyoxal with methylamine. *Geophys. Res. Lett.* **2009**, *36* (L11), L11819.  
Kampf, C. J.; Jakob, R.; Hoffmann, T. Identification and characterization of aging products in the glyoxal/ammonium sulfate system-implications for light-absorbing material in atmospheric aerosols. *Atmos. Chem. Phys.* **2012**, *12* (14), 6323-6333.  
Powelson, M. H.; Espelien, B. M.; Hawkins, L. N.; Galloway, M. M.; De Haan, D. O. Brown carbon formation by aqueous-phase carbonyl compound reactions with amines and ammonium sulfate. *Environ. Sci. Technol.* **2014**, *48* (2), 985-993.  
Aregahegn, K. Z.; Nozière, B.; George, C. Organic aerosol formation photo-enhanced by the formation of secondary photosensitizers in aerosols. *Faraday Discuss.* **2013**, *165*, 123-134.  
González Palacios, L.; Corral Arroyo, P.; Aregahegn, K. Z.; Steimer, S. S.; Bartels-Rausch, T.; Nozière, B.; Ammann, M.; George, C.; and Volkamer, R.: Heterogeneous photochemistry of imidazole-2-carboxaldehyde: HO<sub>2</sub> radical formation and aerosol growth. *Atmos. Chem. Phys. Discuss.* **2016**, *1*, 32.  
National Toxicology, P.: Toxicology and carcinogenesis studies of 4-methylimidazole (Cas No. 822-36-6) in F344/N rats and B6C3F1 mice (feed studies). *Natl. Toxicol. Program Tech Rep Ser.* **2007**, *1*-274.

## Methods

- Atmospheric particles (PM1 or PM10) collected with a DIGITEL DHA-80 high volume filter during 12 or 24 h
- Enrichment from aqueous particle extracts by solid phase extraction (mixed mode, strong cation exchange)
- Analysis by capillary electrophoresis - electrospray ionization - time-of-flight - mass spectrometry (CE-ESI-TOF-MS)

Table 1. Overview of sampling sites, terminology, sampling period, sampling duration and according atmospheric conditions during the sampling period.

Sampling site	Designation	Geographic coordinates	Sampling period (sampling duration per filter)	Comments
Roof of the Leibniz Institute for Tropospheric Research, Leipzig, Germany	TROPOS	51.35 °N, 12.43 °E	January 2014 (12 h, day and night)	Urban background, strong influence of biomass burning aerosol during measurement period, TROPOS and Melpitz are influenced by similar regional air masses, due to proximity
Melpitz, Germany	Melpitz Wi	51.53 °N, 12.93 °E	January to February 2013 (24 h)	Rural background
	Melpitz Su		June to July 2013 (24 h)	
San Pietro Capofiume, Italy	SPC	44.65 °N, 11.62 °E	June 2012 (12 h, day and night)	PEGASOS campaign, rural site
Xianghe, China	Xianghe	39.75 °N, 116.96 °E	July 2013 (12 h, day and night)	CAREBeijing-NCP 2013 campaign, rural background, urban outflow
Wuqing, China	Wuqing	39.39 °N, 117.02 °E	March 2006 (24 h)	HACHI campaign, rural background, urban outflow

## Quantification of imidazoles

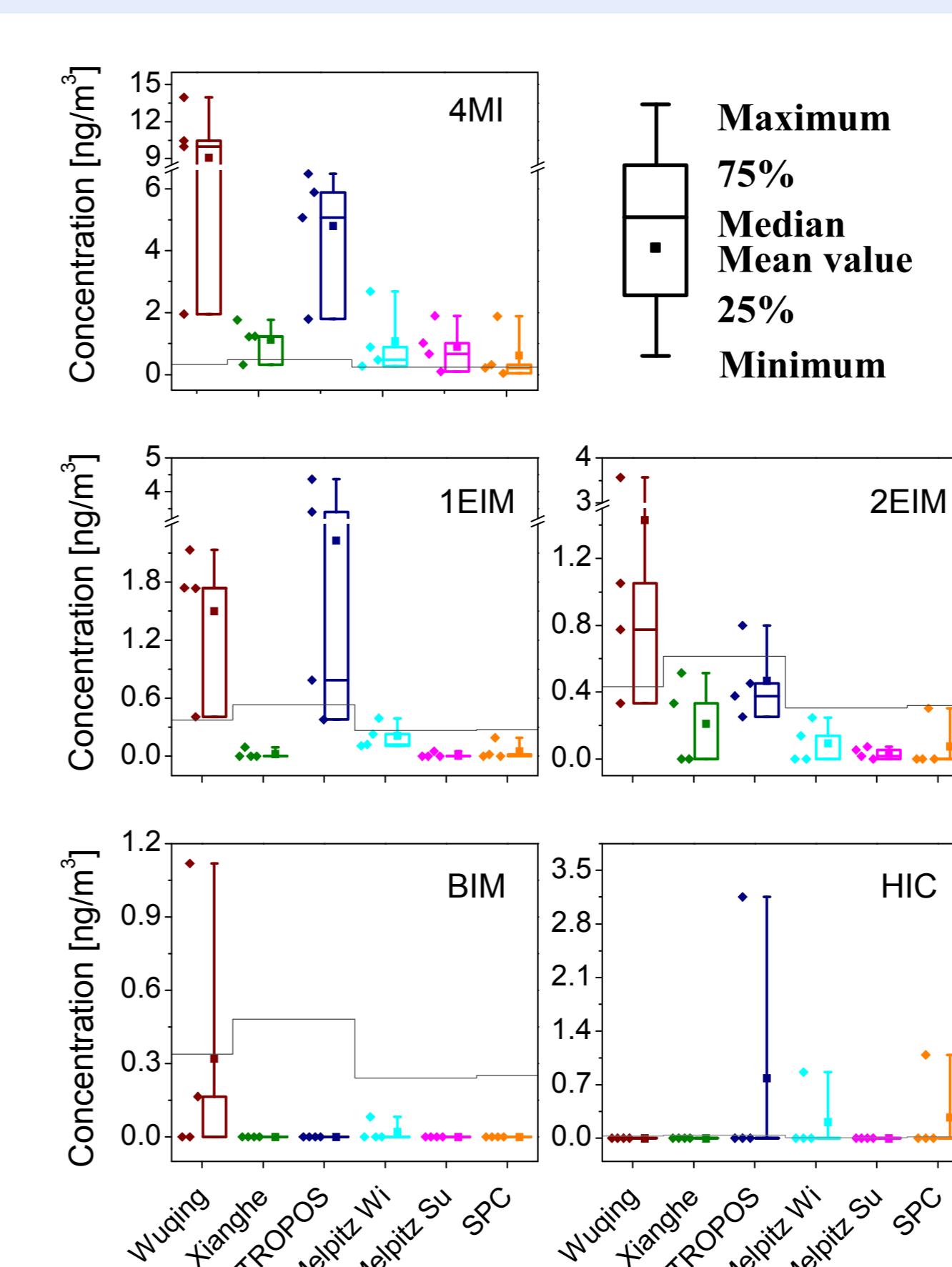


Figure 4. Concentrations of identified imidazoles at the sampling sites plotted as box plots and single data points. The gray solid line represents the mean method detection limit.

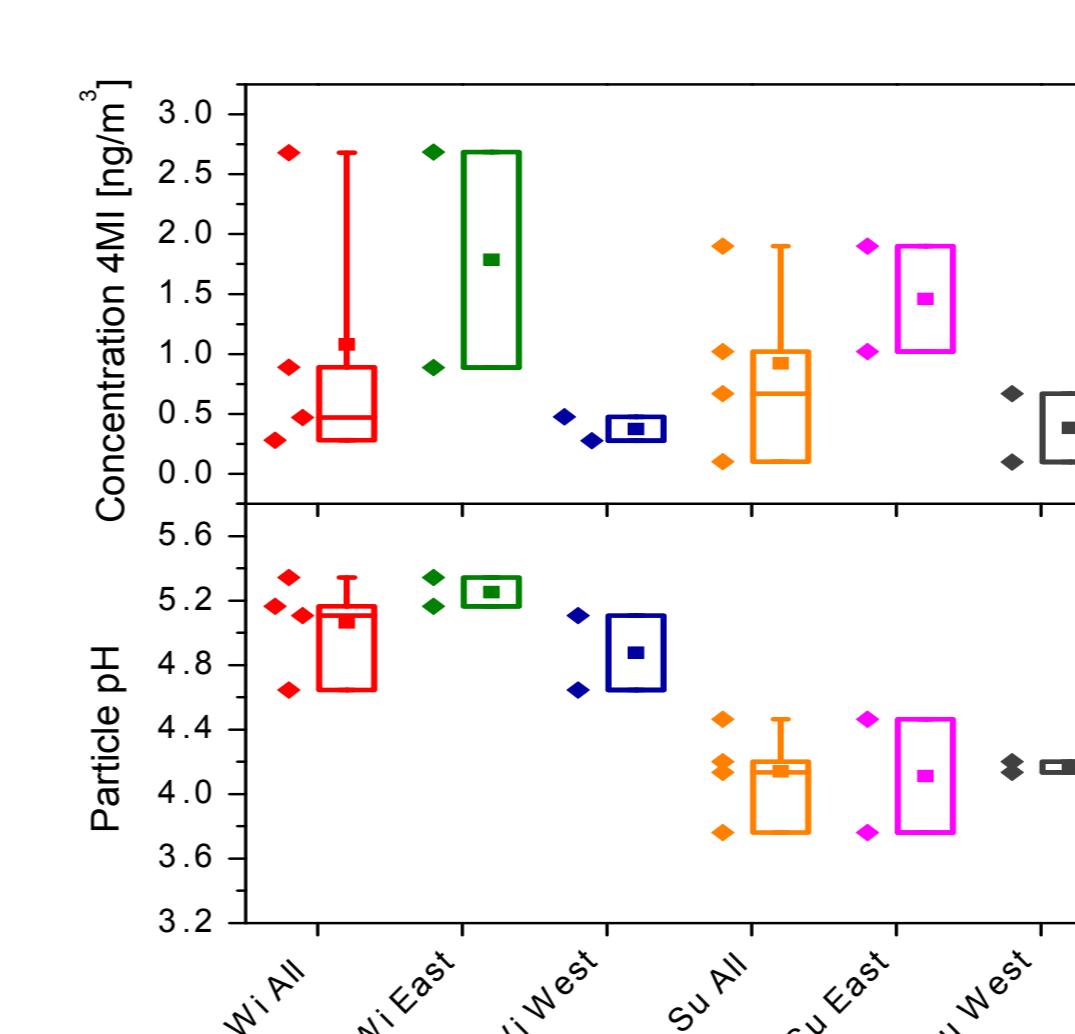


Figure 5. Top: Concentrations of 4MI in summer (Su) and winter (Wi) at Melpitz. Air mass origin is indicated as East and West. Bottom: Aerosol particle pH values at Melpitz.

Screening study:

- Four samples of each campaign (Table 1)
- Concentrations range between 0.2 and 14 ng/m<sup>3</sup>
- 4MI was found to be the most abundant imidazole
- Occurrence of imidazoles seems to be favored at sites with strong biomass burning influence or connected to more polluted air masses
- No connection was found between aerosol particle pH and imidazole concentration
- Concentrations are generally higher at night time than at daytime

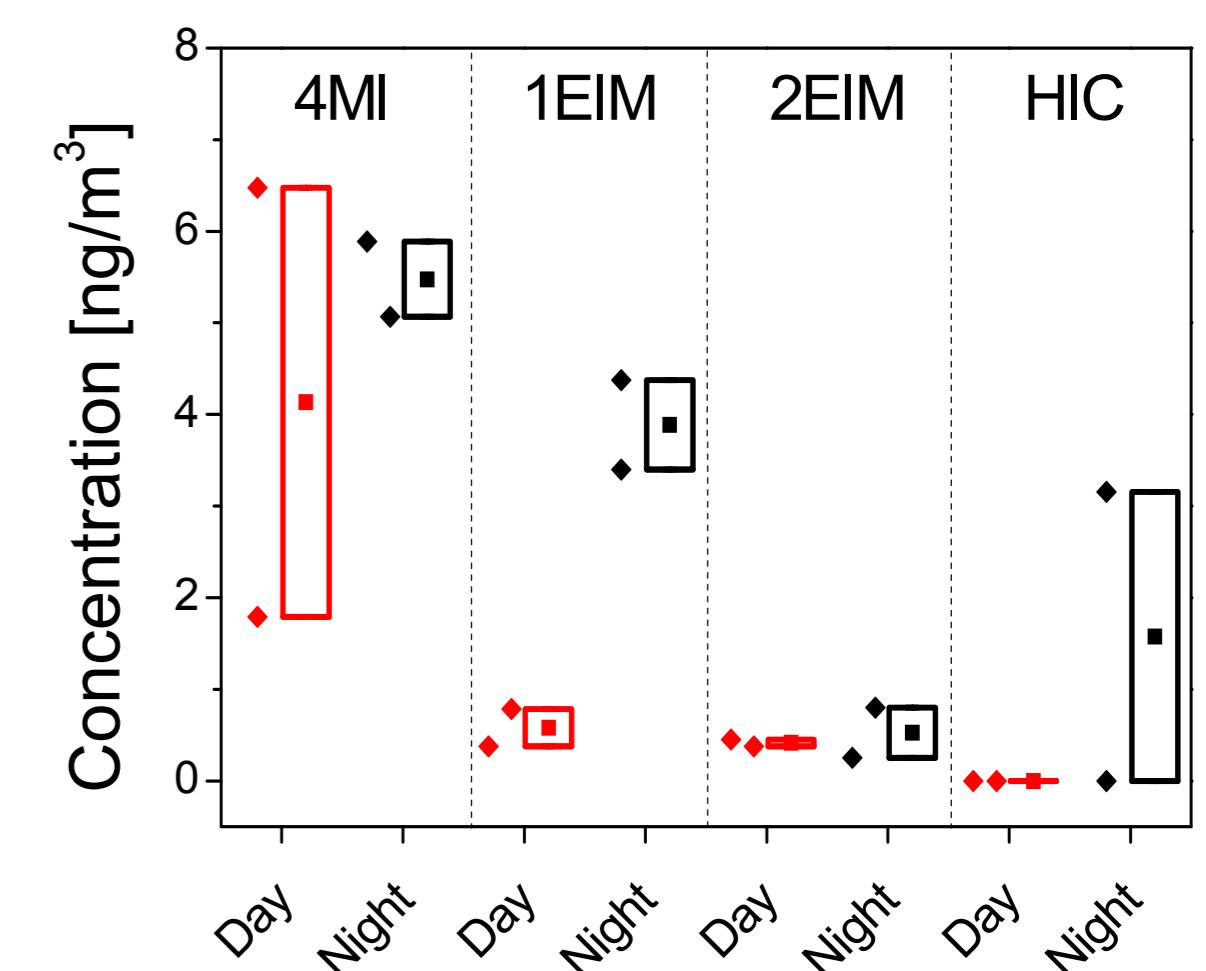


Figure 6. Comparison of daytime and nighttime imidazole concentrations at TROPOS.

## Acknowledgements

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